

Synthesis of some new thienopyrimidines and triazole fused thienopyrimidines and their antimicrobial activities

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ABSTRACT

Some more new derivatives of thienopyrimidines and 1, 3, 4 -triazole fused thienopyrimidines were synthesized efficiently from 2-amino-thieno [2, 3-c] thiochromene-1-carbonitrile **1**. Using the Gewald reaction, the precursor **1** was prepared. The structures of newly synthesized thiophene derivatives were ascertained by spectral and analytical data and they were screened for antimicrobial activity where some of them have showed promising antibacterial and antifungal activities.

Keywords: Thienopyrimidine, Triazolothienopyrimidine, Antibacterial activity, Antifungal activity.

1. INTRODUCTION

Compounds containing the thienopyrimidine nucleus have been prepared by several chemists because of their diverse potent biological properties. [1-6] For instance, a number of thienopyrimidines are known to possess antimalarial, antiallergic, hypo cholesterolemic, analgesic, anti-inflammatory, diuretic, CNS depressant and antiviral activities. [7-9] Literature survey also reveals that triazoles and N-bridged heterocycles fused with them possess diverse pharmacological activities. [10-12] In view of these reports and in continuation of our work on biologically active nitrogen and sulfur heterocycles. [13-21] we have synthesized some more novel triazolothienopyrimidines and evaluated them for their antimicrobial properties. The synthesized compounds were tested against two Gram (+) Bacteria (*staphylococcus aureus*, *Bacillus subtilis*), two Gram (-) bacteria (*Pseudomonas aeruginosa*, *Escherichia coli*) and two yeast-like fungi *Candida albicans* and *Candida parapsilosis* using the broth microdilution method. [22, 23]

2. MATERIALS AND METHODS

Melting points were determined in open capillaries and are uncorrected. The IR spectra were recorded on Nicolet Impact 410 FT IR spectrophotometer using KBr pellets. ¹H and ¹³C NMR were recorded on Bruker 300 MHz FT NMR

spectrometer in CDCl₃ and DMSO-*d*₆ with TMS as internal standard. Mass spectrum was recorded on Finnigan MAT (Model MAT8200) spectrometer and elemental analyses were carried out using Heraeus CHN rapid analyzer.

2.1. Preparation of 2-amino-4H-thieno[2,3-c]thiochromene-1-carbonitrile (**1**)

To a well-stirred mixture of thiochromane-4-one **3** (0.05mol) and malononitrile (3.3g, 0.05mol) in ethanol (45mL) was added elemental sulfur (1.68g, 0.0525mol), diethylamine (5mL) with vigorous stirring during 2 mins. The reaction mixture was stirred at 40-45°C for 1hr. The yellow-orange solid that separated was filtered, washed with hot ethanol and recrystallized from dioxane to yield analytically pure yellow needles. Yield 54%, m.p. 284-286°C; IR (KBr) vcm⁻¹: 3360, 3224, 3033, 2918, 2191, 1629, 1579, 1534; ¹H NMR (300MHz, DMSO-*d*₆) δ: 3.94(s, 2H, S-CH₂), 5.65(s, 2H, NH₂, D₂O exchangeable), 7.01(t, J=7.5Hz, 1H, Ar-H), 7.12(t, J=7.5Hz, 1H, Ar-H), 7.43(d, J=8.1Hz, 1H, Ar-H), 7.68(d, J=8.1Hz, 1H, Ar-H).

2.2. Thiochromene[2,3-c]thieno[2,3-d]pyrimidin-4-one (**2**)

Microwave irradiation of mixture of **1** (1 g) and formic acid (18 mL) at 80° C for 15 minutes yield **2**. The excess of formic acid was removed under reduced pressure. The resulting residue

was crystallized from ethanol to yield pale yellow granules. Yield 78%, m.p. 161-163°C, IR (KBr) ν cm^{-1} : 2934, 1672, 1580, 1364, 974; ^1H NMR (300MHz, CDCl_3) δ : 3.77(s, 2H, S- CH_2), 6.91(t, $J=7.5\text{Hz}$, 1H, Ar-H), 7.11(t, $J=7.5\text{Hz}$, 1H, Ar-H), 7.49(d, $J=8.1\text{Hz}$, 1H, Ar-H), 7.60(d, $J=8.1\text{Hz}$, 1H, Ar-H), 7.86 (s, C2-H, pyrimidine), 11.56 (br s, 1H, NH, D_2O exchangeable).

2.3. 4-Chloro-thiochromene[2,3-c]thieno[2,3-d]pyrimidine (3)

A solution of **2** (2.5 mmol) in dry dioxane (8 mL) was treated with phosphorus oxychloride (1.80 mL) and the mixture was stirred under reflux for 3 h. The reaction mixture was allowed to cool to room temperature and poured into ice-water (200 g), the solid that separated was filtered off and crystallized from petroleum ether. Yield 74%, m.p. 92-94°C; IR cm^{-1} : 3065, 1587, 1323, 965; ^1H NMR (300MHz, CDCl_3) δ : 3.55(s, 2H, S- CH_2), 6.97(t, $J=7.5\text{Hz}$, 1H, Ar-H), 7.23(t, $J=7.5\text{Hz}$, 1H, Ar-H), 7.54(d, $J=8.1\text{Hz}$, 1H, Ar-H), 7.77(d, $J=8.1\text{Hz}$, 1H, Ar-H), 8.01 (s, C2-H, pyrimidine).

2.4. 4-Hydrazino-thiochromene[2,3-c]thieno[2,3-d]pyrimidine (4)

A mixture of compound **3** (2.5 mmol) and hydrazine hydrate 80 % (2 mL) in ethanol (15 mL) was heated at 50°C for 3 h. The reaction mixture was allowed to cool to room temperature. The deposited so precipitate was filtered off and crystallized from dioxane. Yield 72%, m.p. 143-145°C; IR cm^{-1} : 3385, 3320, 3115, 1562, 980; ^1H NMR (300MHz, CDCl_3) δ : 3.72(s, 2H, S- CH_2), 4.56 (br s, 2H, NH_2 , D_2O exchangeable), 5.79 (br s, 1H, NH, D_2O exchangeable), 7.04(t, $J=7.5\text{Hz}$, 1H, Ar-H), 7.20(t, $J=7.5\text{Hz}$, 1H, Ar-H), 7.50(d, $J=8.1\text{Hz}$, 1H, Ar-H), 7.68(d, $J=8.1\text{Hz}$, 1H, Ar-H), 8.08 (s, C2-H, pyrimidine).

2.5. Triazolothienopyrimidine derivative (5a)

A mixture of compound **4** (2 mmol), formic acid (2 mL), and a catalytic amount of concentrated hydrochloric acid was heated under reflux for 7 hours. The reaction mixture was allowed to cool to room temperature and poured into water (200 mL). The precipitate that was formed was collected by filtration, washed with ethanol (50 mL), dried and crystallized from dioxane:ethanol mixture(2:1). Yield 65%, m.p. 143-145°C; IR cm^{-1} : 3100, 2995, 1581, 1360, 961; ^1H NMR (300MHz, CDCl_3) δ : 3.72(s, 2H, S- CH_2), 7.02(t, $J=7.5\text{Hz}$, 1H, Ar-H), 7.18(t, $J=7.5\text{Hz}$, 1H, Ar-H), 7.48(d, $J=8.1\text{Hz}$, 1H, Ar-H), 7.67(d, $J=8.1\text{Hz}$, 1H, Ar-H), 7.91(s, C2-H, pyrimidine), 9.27 (s, 1H, triazole).

2.6. Triazolothienopyrimidine derivative (5b)

A mixture of compound **4** (2 mmol), glacial acetic acid (6 mL) was heated under reflux

for 15 hours. The reaction mixture was allowed to cool to room temperature and poured into water (200 mL). The precipitate that was formed was collected by filtration, dried and crystallized from acetic acid. Yield 68%, m.p. 174-176°C; IR cm^{-1} : 3109, 3018, 1584, 1309, 970; ^1H NMR (300MHz, CDCl_3) δ : 3.72(s, 2H, S- CH_2), 7.02(t, $J=7.5\text{Hz}$, 1H, Ar-H), 7.18(t, $J=7.5\text{Hz}$, 1H, Ar-H), 7.48(d, $J=8.1\text{Hz}$, 1H, Ar-H), 7.67(d, $J=8.1\text{Hz}$, 1H, Ar-H), 7.98 (s, C2-H, pyrimidine).

2.7. Triazolothienopyrimidine derivative (5c)

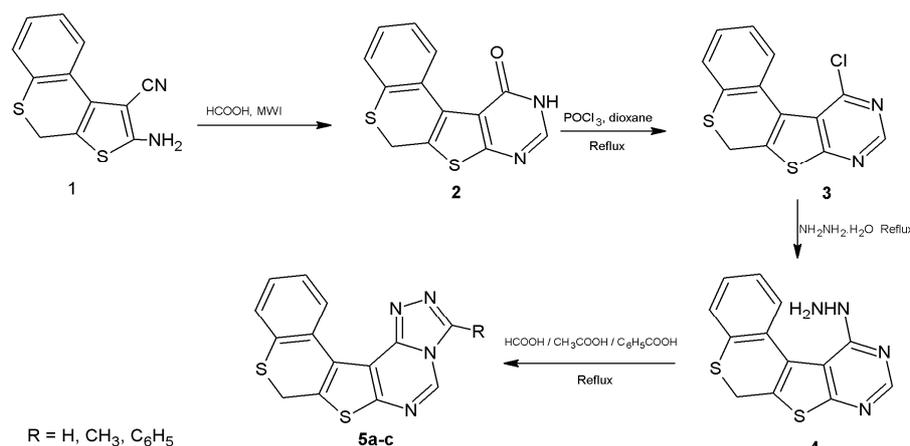
A mixture of compound **4** (2 mmol), Benzoyl chloride (6 mL) was stirred under reflux for 8 hours. The reaction mixture was allowed to cool to room temperature and poured into water (200 mL). The precipitate that was formed was collected by filtration, washed with ethanol (50 mL), dried and crystallized from dioxane:ethanol mixture (2:1). Yield 82%, m.p. 175-177°C; IR cm^{-1} : 3142, 3018, 1560, 1311, 965; ^1H NMR (300MHz, CDCl_3) δ : 3.68(s, 2H, S- CH_2), 7.11(t, $J=7.5\text{Hz}$, 1H, Ar-H), 7.19(t, $J=7.5\text{Hz}$, 1H, Ar-H), 7.52(d, $J=8.1\text{Hz}$, 1H, Ar-H), 7.68(d, $J=8.1\text{Hz}$, 1H, Ar-H), 8.02 (s, C2-H, pyrimidine).

3. RESULTS AND DISCUSSION

As shown in **scheme I**, we employed α -aminocarbonitrile^{15, 16} as a precursor for the synthesis of biologically active thienopyrimidines and triazolothienopyrimidines. 2-amino-thieno [2, 3-c] thiochromene-1-carbo- nitrile **1** was prepared from thiochromene-4-one under conditions reported by K. Gewald.^{17, 18} Formation of thiophene having α -aminonitrile was characterized by the presence of band at 2215 cm^{-1} due to cyano group and N-H stretching bands at 3327 and 3189 cm^{-1} . Further it is also supported by the presence of D_2O exchangeable broad singlet at δ 7.56 in ^1H NMR spectrum due to NH_2 group.

Thienopyrimidin-4-one **2** was prepared by the microwave irradiation of 2-amino-thieno [2, 3-c] thio- chromene-1-carbonitrile **1** in presence of formic acid. The structure of **2** was ascertained by the absence of 2212 cm^{-1} due to cyano group and the presence of $\nu_{\text{C}=\text{O}}$ in IR and a signal at δ 8.26 due to N=CH. And also a D_2O exchangeable broad singlet at δ 11.4 for NH in the ^1H NMR spectrum, along with the other expected signals.

Thienopyrimidin-4-one **2** on treatment with dry dioxane and phosphorous oxychloride afforded the 4-chlorothienopyrimidine derivative **3**. Formation of this product was confirmed by the absence of ν_{NH} and $\nu_{\text{C}=\text{O}}$ bands in IR. Thus obtained 4-chlorothienopyrimidine **3** on treatment with hydrazine hydrate afforded the hydrazino derivative **4**. Formation of the product



Scheme - 1: Synthesis of thienopyrimidines and triazole fused thienopyrimidines.

Table - 1: Antibacterial and antifungal activities of the compounds as MIC values ($\mu\text{g/mL}$)

Compounds	<i>Staphylococcus aureus</i> ATCC 25923	<i>Bacillus subtilis</i> ATCC 6633	<i>Escherichia coli</i> ATCC 25922	<i>Pseudomonas aeruginosa</i> ATCC 27853	<i>Candida albicans</i> ATCC 10231	<i>Candida parapsilosis</i> ATCC 90018
3	512	64	256	256	16	128
4	512	32	256	256	64	64
5a	256	12	128	256	32	64
5b	512	32	256	256	16	128
5c	512	12	256	256	64	256
Ampicillin	4	8	4	-	-	-
Fluconazole	-	-	-	-	8	0.25

was confirmed by the presence of bands at 3429, 3352 and 3238 cm^{-1} in IR spectrum, due to amino functional groups. ^1H NMR spectrum shows D_2O exchangeable singlets at δ 4.86 and 4.39 due to amino groups and the $\text{C}_2\text{-H}$ of pyrimidine resonated at δ 8.20 as a singlet along with other expected signals.

The compound **4** was further converted into triazolothienopyrimidine derivatives (**5a-c**) by treatment with aliphatic acids such as formic or acetic acid or acid chlorides such as benzoyl chloride. The formation of triazole ring involving both amino groups was evident by the absence of absorption bands due to either of these groups in the IR spectrum of **4**. Further ^1H NMR spectrum also exhibited the presence of two characteristic protons each as singlet at δ 8.46 and δ 9.25 due to pyrimidine and triazole proton respectively.

3.1. Antimicrobial activity

Minimum inhibitory concentration (MIC) values for the synthesized compounds were determined by using the broth microdilution method [24, 25]. Two Gram-positive (*S. aureus* ATCC 25923 and *B. subtilis* ATCC 6633) and two Gram-negative (*E. coli* ATCC 25922, *P. aeruginosa*

ATCC 27853) bacteria were used as quality control strains. For determining anti-yeast activities of the compounds, the following reference strains were tested: *Candida albicans* ATCC 10231 and *Candida parapsilosis* ATCC 90018. Ampicillin trihydrate and fluconazole were used as standard antibacterial and antifungal agents, respectively. Fluconazole was dissolved in sterile distilled water, ampicillin trihydrate in phosphate buffer (pH 8) and the stock solution of the synthesized compounds was dissolved in dimethyl sulfoxide (DMSO) and distilled water (50%) at a concentration of 2048 $\mu\text{g/mL}$. Twofold dilutions of the synthesized compounds were prepared (1024, 512... ..2 $\mu\text{g/mL}$), and twofold dilutions of the reference compounds were prepared at 64 – 0.125 $\mu\text{g/mL}$. All bacteria were cultivated in Mueller-Hinton Agar (Merck). The bacteria inoculum was prepared in Mueller-Hinton Broth (Merck) which had been kept at 36 $^\circ\text{C}$ overnight and was diluted with broth to give a final concentration of 5×10^5 cfu/mL. All fungi were cultivated in sabouraud Dextrose Agar (Merck). The fungi inoculum was prepared in sabouraud liquid medium (oxid), which had been kept at 36 $^\circ\text{C}$ overnight and was diluted with RPMI-1640 medium with L-glutamine buffered

with 3-[N-morpholino]-propane sulfonic acid (MOPS) at pH 7 to give a final concentration of 2.5×10^3 cfu/mL. The microplates were incubated at 36°C and read visually after 24 h, except for candida species when it was at 48 h. The incubation chamber was kept humid. At the end of the incubation period, MIC values were recorded as the lowest concentrations of the substances that gave no visible turbidity. The DMSO diluents at a maximum final concentration of 12.5% had no effect on the microorganism's growth. The minimum inhibitory concentrations (MIC) were recorded and as shown in table 1, none of the title compounds had activity against *S. aureus*, *P. aeruginosa* and *E. coli*, but, generally, the title compounds were found to be active against *B. subtilis* and the *fungi*. The antibacterial activity of compounds **5a** and **5c** was 66% of that of ampicillin against *B. subtilis*. Therefore it can be suggested that these compounds show promising antibacterial activity. The antifungal activity of compound **3** and **5b** was 50% of that of fluconazole against *C. albicans*.

4. CONCLUSION

Herein we report the efficient synthesis of some tricyclic and triazole fused tetracyclic thienopyrimidines. All the compounds were characterised by spectral and analytical data and evaluated for their antimicrobial properties. The investigation of antibacterial screening reveals that the compounds **5a** and **5c** have exhibited good antibacterial activity against *B. subtilis* comparable to the standard ampicillin, while **3** and **5b** displayed better antifungal activity against *candida albicans* comparable to the standard fluconazole.

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