International Journal of Chemical and Pharmaceutical Sciences 2014, June., Vol. 5 (2)



Synthesis and characterization of novel 3-substituted-2-butyl-5-chloro-imidazol-4-carbaldehyde derivatives

Namratha B, Nitinkumar S Shetty, Atmesh S Pednekar and Santosh L Gaonkar*.

Department of Chemistry, Manipal Institute of Technolgy, Manipal University, Manipal, Karnataka, India.

* Corresponding Author: E-Mail: gaonkarslg@rediffmail.com

Abstract

2-butyl-5-chloro-4-formyl-3H-imidazole 1 was condensed with the potentially bioactive 5chloromethyl-6-methyl-benzo [1,3] dioxole. Compounds 1 and 2 were subsequently condensed with different aromatic hydrazides to form a novel series of 3-substituted-2-butyl-5-chloroimidazol-4-carbaldehyde derivatives. Structural elucidation was accomplished by ¹H NMR, ¹³C NMR, IR and elemental analyses of synthesized compounds.

Keywords: Hydrazone, Imidazole.

1. INTRODUCTION

Imidazole is an aromatic heterocycle, a diazole and classified as an alkaloid. Synthesis of synthetically [1] imidazoles is and pharmaceutically important as they possess analgesic [2] cardiovascular [3] antiinflammatory^[4], antibacterial ^[5], antifungal ^[6], anticonvolusant^[7], antituberculosis^[8], antiulcer ^[9], and antileishmanicidal activity ^[10]. Moreover, synthesis of substituted imidazoles is in great demand for designing metal-chelating agents [11], corrosion inhibitors, and artificial catalysts [12]. Clotrimazole and Miconazole are the two imidazole drugs which have made their appearances in the market as topical antimycotics. Losartan potassium, an orally active antihypertensive agent ^[13] is a nonpeptide angiotensin II antagonist. 2-butyl-5-chloro-4formyl-3H-imidazole is one of the key intermediates in the synthesis of Losartan [14]. The great potential for different pharmacological activities of imidazole derivatives and their highlighted chemistry has impelled us to synthesize a novel series of 3-substituted-2-butyl-5-chloro-imidazol-4-carbaldehyde derivatives.

2. MATERIALS AND METHODS

2.1. Chemistry protocols

All the experiments were carried out in an Orbit 6 parallel synthesizer. Melting points were determined on a Thomas Hoover melting point apparatus and are uncorrected. ¹H NMR and ¹³C NMR spectra were recorded on a Bruker AM 400 MHz spectrometer using CDCl₃ as solvent and tetramethylsilane as internal standard. Chemical shifts were expressed in δ and the following

abbreviations were used: s = singlet, d = doublet, t = triplet and m = multiplet. IR (KBr) spectra were recorded on Shimadzu 8300 spectrometer. Thin layer chromatography was done with precoated silica gel G plates.

2.1.1. 2-butyl-5-chloro-4-formyl-3H-imidazole 3

A solution of valeronitrile1(5 g, 0.06 mol) in isopropanol (2.166 g, 0.07 mol) and dibutyl ether (5 mL) was cooled to -10 °C. Dry HCl gas was passed over the solution for 2 hour at a temperature below 0 ℃. The mixture was stirred for 48 hour at 0-5℃ and the solid formed was filtered off, washed with chilled ether, and dried to give pentaneimidate hydrochloride, a suspension of which was taken into 15 mL diethyl ether and cooled to -10 °C. Then, slowly added 10 mL of 5 M aqueous NaOH solution and stirred for 15 minutes till the two layers separated. Aqueous layer was extracted with diethyl ether. Ether was evaporated to give crude methyl pentanimidate which was purified by distillation to give pure imidate 2 (2.7 g, 0.02 mol) as oil.

Glycine (1.77 g, 0.02 mol) wascooled to 0-5 $^{\circ}$ in 7 mL of isopropanol and 1 ml water. The pH was adjusted to 8.0-9.0 using 30% KOH solution. A solution of 2.7 g imidate in 4.86 mL toluene was added and the mixture was stirred at room temperature for 12 hour with the pH adjusted to 7 using Conc. H₂SO₄.The resulting suspension was refluxed at 80 $^{\circ}$ upon addition of POCl₃(10.26 g, 0.06 mol) anddimethyl formamide(4.806 g, 0.06 mol) for 2 h, cooled to room temperature, poured onto ice water. This mixture was extracted into toluene and washed with water. The organic phase was concentrated to get a white crystalline solid of 2-Butyl-5chloro-4-formyl-3H-imidazole **3**(2.26 g, 0.011 mol).Melting point (m.p.) was 98-100 °C.¹H NMR CDCl₃: δ 1.11 (t, *J* =1.0 Hz, 3H, CH₃), 1.49 (m, 2H, CH₂), 2.25 (m, 2H, CH₂), 2.85 (t, *J* =1.3 Hz, 2H, CH₂), 9.56 (s, 1H, CHO), 13.28 (bs, 1H, NH). ¹³C NMR CDCl₃: δ 13.2 (q), 23.2 (t), 28.1 (t), 30.1 (t), 128.2 (s), 144.1 (s), 158.8 (s), 179.2 (d). IR (KBr pellets cm⁻¹) *v* 3409, 3070, 2968, 2829, 1671, 1459. Anal.Calcd.for C₈H₁₁ClN₂O: C, 51.48; H, 5.94; N, 15.01 %. Found: C, 51.38; H, 5.98; N, 15.10 %.

2.1.2. 2-butyl-5-chloro-4-formyl-3-(6-methylbenzo[1,3]dioxol-5-ylmethyl)imidazole 4

A parallel synthesis equipment was charged 2-butyl-5-chloro-4-formyl-3Hwith imidazole 3 (1 g, 5.37 mmol), anhydrous K₂CO₃ (0.90 g, 6.52 mmol), 5-chloromethyl-6-methylbenzo [1,3] dioxole (0.98 g, 5.32 mmol) and dimethyl formamide (5 mL). The reaction mixture was stirred overnight at room temperature. After completion of the reaction as marked by TLC (toluene: ethyl acetate = 7: 3), the dark red mass was diluted with 25 mL water and the product was extracted with dichloromethane (25 mL). The extract was washed with water (10 mL) and dried (Na₂SO₄). The solvent was evaporated and the remaining pale yellow oil was crystallized from ethanol as a white crystalline solid 4 with a yield of (1.48g, 83 %), m.p. 110-112 °C.¹H NMR CDCl₃: δ 0.79 (t, J = 1.0 Hz, 3H, CH₃), 1.33 (m, 2H, CH₂), 1.79 (m, 2H, CH₂), 2.49 (s, 3H, CH₃), 2.57 (t, / =1.3 Hz, 2H, CH₂), 5.41 (s, 2H, CH₂), 5.90 (s, 2H, CH₂), 6.82 (s, 2H, ArH), 9.62 (s, 1H, CHO). ¹³C NMR CDCl₃: δ 13.8 (q), 23.1 (t), 23.9 (q), 28.2 (t), 30.5 (t), 34.9 (t), 91.5 (t), 113.2 (d), 123.3 (d), 128.1 (s), 130.2 (s), 131.4 (s), 140.1 (s), 141.7 (s), 144.3 (s), 159.1 (s), 189.2 (d). IR (KBr pellets cm⁻¹) v 3071, 2961, 2859, 1763, 1667, 1452, 1230. Anal.Calcd.for C₁₇H₁₉ClN₂O₂: C, 60.99; H, 5.72; N, 8.37 %. Found: C, 60.95; H, 5.79; N, 8.42 %.

2.1.3. Representative procedure for the synthesis of aroylhydrazones 5 and 6

0.1 g of 3 or 4 was taken in parallel synthesis equipment, equipped with magnetic stirrer. Added an equimolar proportion of aromatic hydrazides, isopropanol (2 mL) and refluxed for 2h. The progress of the reaction was monitored by TLC (toluene: ethyl acetate: diethylamine = 7.5: 2.5:1). After completion of the reaction, the mass was cooled and the solid formed was filtered to give 5or 6respectively.

The same procedure was used in all cases.

2.1.3.1. [(2-butyl-5-chloro-3H-imidazol-4yl)methylidene]-4-nitrobenzohydrazone 5a Obtained from 3 (0.1 g, 0.53 mmol) and 4nitrobenzohydrazide (0.099 g, 0.53 mmol) as a yellow crystalline solid (0.161 g, 87 %), m.p. 158-160 °C.¹H NMR CDCl₃: δ 0.877 (t, *J* =1.0 Hz, 3H, CH₃), 1.28 (m, 2H, CH₂), 1.61 (m, 2H, CH₂), 2.61(t, *J* =1.3 Hz, 2H, CH₂), 7.77 (s, 1H, CH=N-), 8.17 (s, 2H, ArH), 8.34 (s, 2H, ArH), 11.09 (bs, 1H, CONHN), 12.09 (bs, 1H, NH). ¹³C NMR CDCl₃: δ 13.1 (q), 23.2 (t), 28.1 (t), 31.3 (t), 2×122.2 (d), 128.4 (s), 2×130.1 (d), 135.5 (s), 143.9 (s), 147.6 (d), 151.8 (s), 157.2 (s), 168.8 (s). IR (KBr pellets cm⁻¹) *v* 3440, 1666, 1620, 1519, 1342, 709.Anal.Calcd.for C₁₅H₁₆ClN₅O₃: C, 51.36; H, 4.88; N, 19.96 %. Found: C, 51.30; H, 4.90; N, 19.92 %.

2.1.3.2. [(2-butyl-5-chloro-3H-imidazol-4yl)methylidene]-4-bromobenzohydrazone 5b

Obtained from 3 (0.1 g, 0.53 mmol) and 4-bromobenzohydrazide (0.115 g, 0.53 mmol) as a yellow crystalline solid (0.168 g, 83 %), m.p. 106-107 °C.¹H NMR CDCl₃: δ 0.876 (t, *J* =1.0 Hz, 3H, CH₃), 1.28 (m, 2H, CH₂), 1.61 (m, 2H, CH₂), 2.63 (t, *J* =1.3 Hz, 2H, CH₂), 7.41 (s, 2H, ArH), 7.62 (s, 2H, ArH), 7.77 (s, 1H, CH=N-), 11.06 (bs, 1H, CONHN), 12.08 (bs, 1H, NH). ¹³C NMR CDCl₃: δ 13.3 (q), 23.2 (t), 28.6 (t), 30.1 (t), 126.9 (s), 127.2 (s), 128.7 (s), 2×129.4 (d), 2×131.7 (d), 144.3 (s), 147.5 (d), 158.2 (s), 168.8 (s). IR (KBr pellets cm⁻¹) *v* 3286, 3085, 1658, 1620, 1550, 524.Anal.Calcd.for C₁₅H₁₆BrClN₄O: C, 46.83; H, 4.45; N, 14.56 %. Found: C, 46.87; H, 4.43; N, 14.59 %.

2.1.3.3. [(2-butyl-5-chloro-3H-imidazol-4-yl)methylidene]-4-chlorobenzohydrazone 5c

Obtained from **3** (0.1 g, 0.53 mmol) and 4chlorobenzohydrazide (0.091 g, 0.53 mmol) as a yellow crystalline solid (0.152 g, 85 %), m.p. 114-116 °C.¹H NMR CDCl₃: δ 0.878 (t, *J* =1.0 Hz, 3H, CH₃), 1.25 (m, 2H, CH₂), 1.61 (m, 2H, CH₂), 2.61 (t, *J* =1.3 Hz, 2H, CH₂), 7.38 (s, 2H, ArH), 7.51 (s, 2H, ArH), 7.79 (s, 1H, CH=N-), 11.09 (bs, 1H, CONHN), 12.09 (bs, 1H, NH). ¹³C NMR CDCl₃: δ 13.2 (q), 23.3 (t), 28.9 (t), 30.0 (t), 2× 126.7 (d), 127.6 (s), 128.5 (s), 2×130.4 (d), 137.6 (s), 144.1 (s), 147.5 (d), 157.9 (s), 168.2 (s). IR (KBr pellets cm⁻¹) *v* 3471, 3178, 1635, 1558, 1496, 840, 740.Anal.Calcd.for C₁₅H₁₆Cl₂N₄O: C,

2.1.3.4. [(2-butyl-5-chloro-3H-imidazol-4-yl)methylidene]-4-methylbenzohydrazone 5d

Obtained from 3 (0.1 g, 0.53 mmol) and 4methylbenzohydrazide (0.08 g, 0.53 mmol) as a yellow crystalline solid (0.15 g, 89 %), m.p. 126-128 °C.¹H NMR CDCl₃: δ 0.876 (t, *J* =1.0 Hz, 3H, CH₃), 1.28 (m, 2H, CH₂), 2.18 (m, 2H, CH₂), 2.49 (s, 3H, CH₃), 2.61 (t, *J* =1.3 Hz, 2H, CH₂), 7.39 (s, 2H, ArH), 7.58 (s, 2H, ArH), 8.32 (s, 1H, CH=N-), 11.06 (bs, 1H, CONHN), 12.08 (bs, 1H, NH). ¹³C NMR CDCl₃: δ 13.2 (q), 21.1 (q), 23.2 (t), 28.1 (t), 30.1 (t), 2×126.9 (d), 127.7 (s), 128.5 (s), 2×129.4 (d), 140.6 (s), 144.1 (s), 147.2 (s), 158.8 (s), 168.6 (s). IR (KBr pellets cm⁻¹) v 3456, 3078-3039, 2947, 1643, 1566, 709.Anal.Calcd.for C₁₆H₁₉ClN₄O: C, 60.09; H, 6.30; N, 17.52 %. Found: C, 60.01; H, 6.36; N, 17.58 %.

2.1.3.5. [(2-butyl-5-chloro-3H-imidazol-4-yl)methylidene]pyridine-3-carbohydrazone 5e

Obtained from 3 (0.1 g, 0.53 mmol) and pyridine-3-carbohydrazide (0.073 g, 0.53 mmol) as a yellow crystalline solid (0.128 g, 79 %), m.p. 234-236 °C.¹H NMR CDCl₃: δ 0.873 (t, *J* =1.0 Hz, 3H, CH₃), 1.21 (m, 2H, CH₂), 1.61 (m, 2H, CH₂), 2.61 (t, *J* =1.3 Hz, 2H, CH₂), 7.54 (s, 1H, CH=N-), 7.5-9.0 (m, 4H, PyH), 11.86 (bs, 1H, CONHN), 12.90 (bs, 1H, NH). ¹³C NMR CDCl₃: δ 13.3 (q), 23.4 (t), 28.4 (t), 30.1 (t), 122.9 (d), 124.1 (s), 128.2 (s), 135.9 (d), 144.1 (s), 147.5 (d), 151.3 (d), 154.9 (s), 158.7 (s), 168.6 (s). IR (KBr pellets cm⁻¹) *v* 3286, 3224, 3062, 1650, 1604, 1542, 1049, 702.Anal.Calcd.for C₁₄H₁₆ClN₅O: C, 55.81; H, 5.59; N, 16.47 %. Found: C, 54.85; H, 5.52; N, 16.41 %.

2.1.3.6. [(2-butyl-5-chloro-3H-imidazol-4-yl)methylidene]-2-hydroxybenzohydrazone 5f

Obtained from 3 (0.1 g, 0.53 mmol) and 2hydroxybenzohydrazide (0.084 g, 0.53 mmol) as a white crystalline solid (0.139 g, 82 %), m.p. 110-112 °C.¹H NMR CDCl₃: δ 1.32 (t, *J* =1.0 Hz, 3H, CH₃), 1.56 (m, 2H, CH₂), 1.88 (m, 2H, CH₂), 2.56 (t, *J* =1.3 Hz, 2H, CH₂), 6.9-7.7 (m, 4H, ArH), 7.87 (s, 1H, CH=N-), 8.35 (s, 1H, OH), 11.75 (bs, 1H, CONHN), 11.93 (bs, 1H, NH). ¹³C NMR CDCl₃: δ 13.7 (q), 23.2 (t), 28.1 (t), 30.3 (t), 113.8 (d), 115.2 (s), 120.5 (d), 128.3 (s), 129.7 (d), 133.1 (d), 144.2 (s), 147.4 (d), 156.1 (s), 158.4 (s), 169.1 (s). IR (KBr pellets cm⁻¹) *v* 3471, 3263, 3018, 1635, 1542, 848, 748.Anal.Calcd.for C₁₅H₁₇ClN₄O₂: C, 55.99; H, 5.64; N, 17.41 %. Found: C, 55.92; H, 5.69; N, 17.45 %.

2.1.3.7. [2-butyl-5-chloro-4-formyl-3-(6methyl-benzo[1,3]dioxol-5-ylmethyl)imidazol-4-yl- methylidine]-4-nitrobenzohydrazone 6a

Obtained from 4 (0.1 g, 0.28 mmol) and 4nitrobenzohydrazide (0.053 g, 0.28 mmol) as a vellow crystalline solid (0.124 g, 86 %), m.p. 108-109 °C.¹H NMR CDCl₃: δ 0.92 (t, *J* =1.0 Hz, 3H, CH₃), 1.39 (m, 2H, CH₂), 1.56 (m, 2H, CH₂), 2.27 (s, 3H, CH₃), 2.59 (t, J = 1.3 Hz, 2H, CH₂), 4.91 (s, 2H, CH₂), 5.88 (s, 2H, CH₂), 6.43 (s, 2H, ArH), 7.72 (s, 1H, CH=N-), 8.14 (s, 2H, ArH), 8.51 (s, 2H, ArH), 10.09 (bs, 1H, CONHN). ¹³C NMR CDCl₃: δ 13.8 (q), 23.1 (t), 23.9 (q), 28.4 (t), 30.5 (t), 34.9 (t), 91.6 (t), 113.2 (d), 2×122.1(d), 123.3 (d), 128.1 (s), 2×130.1 (d), 130.3 (s), 131.6 (s), 135.3 (s), 140.1 (s), 141.7 (s), 143.9 (s), 144.3 (s), 147.6 (d), 159.3 (s), 168.5 (s) . IR (KBr pellets cm⁻¹) v 3371, 3031, 1612, 1542, 1342, 1033.Anal.Calcd.for C₂₄H₂₄ClN₅O₅: C, 57.77; H, 5.05; N, 14.04 %. Found: C, 57.79; H, 5.00; N, 14.08 %.

2.1.3.8. [2-butyl-5-chloro-4-formyl-3-(6methyl-benzo[1,3]dioxol-5-ylmethyl)imidazol-4-yl- methylidine]-4-bromobenzohydrazone 6b

Obtained from 4 (0.1 g, 0.28 mmol) and 4bromobenzohydrazide (0.06 g, 0.28 mmol) as a vellow crystalline solid (0.123 g, 80 %), m.p. 178-179 °C.¹H NMR CDCl₃: δ 0.83 (t. *I* = 1.0 Hz. 3H, CH₃). 1.30 (m, 2H, CH₂), 1.56 (m, 2H, CH₂), 2.29 (s, 3H, CH₃), 2.55 (t, *J* = 1.3 Hz, 2H, CH₂), 5.67 (s, 2H, CH₂), 5.91 (s, 2H, CH₂), 6.40 (s, 2H, ArH), 7.72 (s, 2H, ArH), 7.81 (s, 2H, ArH), 8.32 (s, 1H, CH=N-), 11.77 (bs, 1H, CONHN). ¹³C NMR CDCl₃: δ 14.05 (q), 18.93 (t), 22.12 (q), 26.0 (t), 29.18 (t), 46.45 (t), 91.6 (t), 113.2 (d), 123.4 (d), 126.8 (s), 127.2 (s), 128.4 (s), 2×129.8 (s), 130.2 (s), 131.4 (s), 2×131.7 (d), 140.1 (s), 141.2 (s), 144.3 (s), 147.6 (d), 152.45 (s), 162.01 (s). IR (KBr pellets cm⁻¹) v 3193, 3047, 1650, 1558, 1041, 671.Anal.Calcd.for C₂₄H₂₄ BrClN₄O₃: C, 54.10; H, 4.73; N, 10.51 %. Found: C, 54.15; H, 4.77; N, 10.59 %.

2.1.3.9. [2-butyl-5-chloro-4-formyl-3-(6methyl-benzo[1,3]dioxol-5-ylmethyl)imidazol-4-yl- methylidine]-4-chlorobenzohydrazone 6c

Obtained from 4 (0.1 g, 0.28 mmol) and 4-chlorobenzohydrazide (0.048 g, 0.28 mmol) as a vellow crystalline solid (0.118 g, 84 %), m.p. 120-121 °C.¹H NMR CDCl₃: δ 0.91 (t, *J* = 1.0 Hz, 3H, CH₃), 1.39 (m, 2H, CH₂), 1.55 (m, 2H, CH₂), 2.27 (s, 3H, CH₃), 2.56 (t, J = 1.3 Hz, 2H, CH₂), 4.91 (s, 2H, CH₂), 5.93 (s, 2H, CH₂), 6.44 (s, 2H, ArH), 7.39 (s, 2H, ArH), 7.50 (s, 2H, ArH), 7.76 (s, 1H, CH=N-), 10.06 (bs, 1H, CONHN). ¹³C NMR CDCl₃: δ 13.7 (q), 23.1 (t), 23.8 (q), 28.2 (t), 30.5 (t), 34.9 (t), 91.6 (t), 113.8 (d), 123.4 (d), 2×126.5 (d), 127.5 (s), 128.5 (s), 130.1 (s), 2×130.7 (d), 131.6 (s), 137.6 (s), 140.2 (s), 141.6 (s), 144.1 (s), 147.6 (d), 159.2 (s), 168.2 (s). IR (KBr pellets cm⁻¹) v 3201, 2954, 1666, 1558, 1157, 1041.Anal.Calcd.for C₂₄H₂₄Cl₂N₄O₃: C, 59.02; H, 5.16; N, 11.47 %. Found: C, 59.08; H, 5.11; N, 11.45 %.

2.1.3.10 [2-butyl-5-chloro-4-formyl-3-(6methyl-benzo[1,3]dioxol-5-ylmethyl)imidazol-4-yl- methylidine]-4-methylbenzohydrazone 6d

Obtained from 4 (0.1 g, 0.28 mmol) and 4methylbenzohydrazide (0.042 g, 0.28 mmol) as a yellow crystalline solid (0.112 g, 87 %), m.p. 162-163 °C.¹H NMR CDCl₃: δ 0.94 (t, *J* =1.0 Hz, 3H, CH₃), 1.35 (m, 2H, CH₂), 1.59 (m, 2H, CH₂), 2.29 (s, 3H, CH₃), 2.37 (s, 3H, CH₃), 2.59 (t, *J* =1.3 Hz, 2H, CH₂), 4.97 (s, 2H, CH₂), 5.91 (s, 2H, CH₂), 6.41 (s, 2H, ArH), 7.38 (s, 2H, ArH), 7.56 (s, 2H, ArH), 7.76 (s, 1H, CH=N-), 10.01 (bs, 1H, CONHN). ¹³C NMR CDCl₃: δ 13.6 (q), 23.1 (q), 23.2 (t), 23.9 (q), 28.1 (t), 30.2 (t), 34.8 (t), 91.1 (t), 113.1 (d), 123.2 (d), 2×126.3 (d), 127.5 (s), 128.2 (s), 2×129.3 (d), 130.2 (s), 131.4 (s), 140.1 (s), 140.9 (s), 141.7 (s), 144.3 (s), 147.2 (d), 159.1 (s), 168.6 (s). IR (KBr pellets cm⁻¹) *v* 3039, 2947, 1650, 1488, 1041.Anal.Calcd.for C₂₅H₂₇ClN₄O₃: C, 64.16; H, 6.03; N, 11.97 %. Found: C, 64.12; H, 6.07; N, 11.90 %.

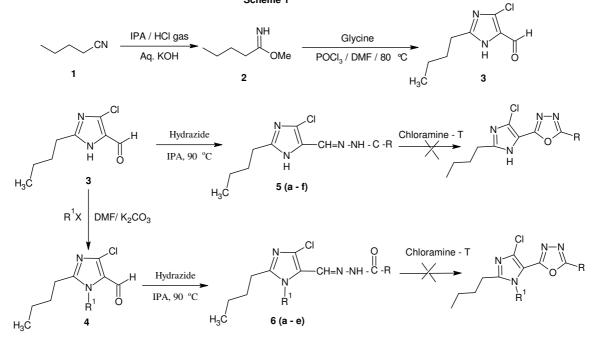
2.1.3.11. [2-butyl-5-chloro-4-formyl-3-(6methyl-benzo[1,3]dioxol-5-ylmethyl)imidazol-4-yl-methylidine]-pyridine-3-carbohydrazone 6e

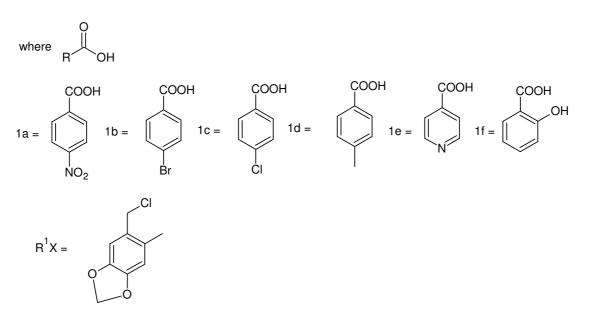
Obtained from 4 (0.1 g, 0.28 mmol) and pyridine-3-carbohydrazide (0.038 g, 0.28 mmol) as a vellow crystalline solid (0.108 g, 81 %), m.p. 117-118 °C.¹H NMR CDCl₃: δ 0.92 (t, J =1.0 Hz, 3H, CH₃), 1.37 (m, 2H, CH₂), 1.56 (m, 2H, CH₂), 2.29 (s, 3H, CH₃), 2.56 (t, J =1.3 Hz, 2H, CH₂), 4.92 (s, 2H, CH₂), 5.91 (s, 2H, CH₂), 6.42 (s, 2H, ArH), 7.75 (s, 1H, CH=N-), 8.0-8.9 (m, 4H, PyH), 10.06 (bs, 1H, CONHN). ¹³C NMR CDCl₃: δ 13.8 (q), 23.2 (t), 23.9 (q), 28.2 (t), 30.5 (t), 34.9 (t), 91.7 (t), 113.4 (d), 122.8 (d), 123.2 (d), 124.1 (s), 128.6 (s), 130.1 (s), 131.7 (s), 135.8 (d), 140.1 (s), 141.9 (s), 144.2 (s), 147.5 (d), 151.3 (d), 154.8 (d), 159.2 (s), 168.9 (s). IR (KBr pellets cm⁻¹) v 3232, 3055, 1612, 1558, 1041, 709.Anal.Calcd.for C₂₃H₂₄ClN₅O₃: C, 60.72; H, 5.54; N, 15.39 %. Found: C, 60.75; H, 5.56; N, 15.38 %.

3. RESULTS AND DISCUSSION

Compounds were synthesized as per scheme 1. ¹H NMR, ¹³C NMR, IR and elemental analyses characterized all the synthesized compounds. 2-butyl-5-chloro-4-formyl-3H-Scheme 1

imidazole 3 was prepared by the reaction between imidate and glycine. ¹H NMR spectrum of this aldehyde showed a broad singlet at δ 13.28 due to NH group of imidazole ring and an aldehydic proton at δ 9.56. The aldehydic carbon was observed at δ 179.2. 2-butyl-5-chloro-4-formyl-3-(6-methyl-benzo[1,3] dioxol-5-ylmethyl) imidazole 4was synthesized by condensing compound 3with 5 -chloromethyl-6-methyl-benzo [1,3] dioxoleR¹X. 2-butyl-5-chloro-4-formyl-3-(6methyl-benzo[1,3] dioxol-5-ylmethyl) imidazole has proven to show antiinflammatory activity [4]. Compound4showed the aldehydic proton at δ 9.62 and aldehydic carbon at δ 189.2. Disappearance of peak at δ 13.28 due to NH group of imidazole ringand appearance of a new peak at δ 5.41 due to NCH₂ group indicated the formation of compound4. All other substituents were observed in the expected region.[(2-butyl-5-chloro-3Himidazol-4-yl)methylidene]-carbohydrazones5(aand [2-butyl-5-chloro-4-formyl-3-(6-methylf benzo[1,3]dioxol-5-ylmethyl)imidazol-4-ylmethylidine]carbohydrazones6(a-e) were synthesized by the condensation of 3 and 4with hydrazidesrespectively. ¹H NMR spectra of synthesized aryl hydrazones showed the disappearance of the aldehydic proton at δ 9.5-9.6 and appearance of a singlet at δ 7.75-9.06 due to CH=N- group and a broad singlet at δ 10.02-12.09 due to CONHN group. Furthermore, ¹³C NMR spectra showed the disappearance of the aldehydic carbon at δ 179-189 and appearance of a doublet at δ 147.1-147.8 due to CH=N- group and a singlet at δ 168-169 due to CONHN group. All the other substituents were observed in the expected regions.





We have also attempted to synthesize 2-(2-butyl-5-chloro-3-substituted-imidazol-4-yl)-5substituted-1,3,4-oxadiazoles from aryl hydrazones5(a-f) and 6(a-e) using mild oxidant Chloramine-T. Chloramine – T is preferred over the classical oxidants like acetic anhydride and KMnO₄ to avoid side reactions.But, the reaction failed to afford the desired products. This may be due to the presence of bulky -Cl group attached to C-5 of 3and4 which sterically hinders the oxidative cyclization pathway.

4. CONCLUSION

A novel series of [(2-butyl-5-chloro-3-substituted-imidazol-4-

yl)methylidene]carbohydrazones were synthesized and characterized based on their physical and spectral data. This work rules out the formation of 2-(2-butyl-5-chloro-3-substitutedimidazol-4-yl)-5-substituted-1,3,4-oxadiazoles form the synthesized hydrazones owing to the steric hindrance in *N*-substituted 2-butyl-5chloro-4-formyl-3H-imidazole.

Acknowledgements

Manipal Institute of Technology, Manipal, is gratefully acknowledged for providing the laboratory facilities to carry out the research work.

5. REFERENCES

- Gaonkar SL and Lokanatha Rai KM. 2-Butyl-5chloro-3H-imidazole-4-carbaldehyde as a new synthon for the synthesis of fused ring heterocycles via intramolecular 1,3-Dipolar cycloaddition reactions. Journal of Heterocyclic Chemistry.2010; 47: 543-546.
- 2. Umit U, Nalan G and KaraburunIhan. Synthesis and analgesic activity of some 1-

benzyl-2-substituted-4, 5-diphenyl-1Himidazolederivatives. **11Farmaco**, 2001; 56: 285-290.

- 3. Robertson BW, Beedle EE and Krushinski JH. Structure-activity relationships of arylimidazopyridinecardiotonics: discovery and inotropic activity of 2-[2-methoxy-4-(methylsulfinyl)phenyl]-1H-imidazo[4,5c]pyridine. Journal of Medicinal Chemistry, 1985; 28: 717-727.
- Gaonkar SL, LokanathaRai KM and SuchethaShetty N. Microwave-assisted synthesis and evaluation of antiinflammatory activity of new series of N-substituted 2-butyl-5-chloro-3H-imidazolederivatives. Medicinal Chemistry Research, 2008; 18: 221-230.
- Slee DH, Romano SJ, Yu J, Nguyen TN. Development of potent non-carbohydrate imidazole-based small molecule selectin inhibitors with antiinflammatory activity. Journal of Medicinal Chemistry, 2001; 44: 2094-2107.
- Khabandideh S, Rezaei Z, Bahrinajafi Ret al.Synthesis of N-alkylated derivatives of imidazole as antibacterial agents. Biooranic & Medicinal Chemistry Letters, 2003; 13: 2863-2865.
- Gunay NS, Ulusoy GN, Ergenc N and Otuk G. 5-Nitroimidazole derivatives as possible antibacterial and antifungal agents. Farmaco, 1999; 54: 826-831.
- 8. Soyer Z, Sultan F, Erol KK and Pabuccuolu V. Synthesis and anticonvulsant activity of some x-(1Himidazol-1-yl)-N-phenylacetamide and

propionamide derivatives. **Farmaco**, 2004; 59: 595-600.

- 9. Gupta P, Hameed S, Jain R. Ring-substituted imidazoles as a new class of antituberculosis agents. **European Journal of Medicinal Chemistry**, 2004; 39: 805-814.
- 10. Johnson RA, Huong SM and Huang ES. Activation of the mitogen-activated protein kinase p38 by human cytomegalovirus infection through two distinct pathways: a novel mechanism for activation of p38. Journal of Virology, 2000; 74: 1158-1167.
- 11. Vanelle P, Meuche J and Crozet MP. Functional derivatives of 5-benzo[1, 3]dioxol-5-yl-1methyl-1H-imidazole-2-carbaldehyde and evaluation of leishmanicidal activity. **European Journal of Medicinal Chemistry**.2000; 35: 157-162.
- Knapp S, Albaneze J, Schugar HJ. The beneficial effect of imidazole ligands on (pi.allyl) nickel Coupling. Journal of Organic Chemistry, 1993; 58: 997-998.
- 13. Benner SA and Heeb NV. Guanosine derivatives bearing an N 2-3imidazolepropionic acid. **Tetrahedron** Letters.1994; 35: 3045-3048.
- 14. Carini DJ, Duncia JV, Aldrich PE and Chiu AT. Nonpeptide angiotensin II receptor antagonists: the discovery of a series of N-(biphenylylmethyl) imidazoles as potent, orally active antihypertensives.**Journal of Medicinal Chemistry**, 1991; 34: 2525-2547.